# Nonpeptide Urotensin-II Receptor Antagonists: A New Ligand Class Based on Piperazino-Phthalimide and Piperazino-Isoindolinone Subunits

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### Received May 21, 2009

We have discovered two related chemical series of nonpeptide urotensin-II (U-II) receptor antagonists based on piperazino-phthalimide (5 and 6) and piperazino-isoindolinone (7) scaffolds. These structure types are distinctive from those of U-II receptor antagonist series reported in the literature. Antagonist 7a exhibited single-digit nanomolar potency in rat and human cell-based functional assays, as well as strong binding to the human U-II receptor. In advanced pharmacological testing, 7a blocked the effects of U-II in vitro in a rat aortic ring assay and in vivo in a rat ear-flush model. A discussion of U-II receptor antagonist pharmacophores is presented, and a specifically defined model is suggested from tricycle 13, which has a high degree of conformational constraint.

## Introduction

The cyclic peptide urotensin-II  $(U-H<sup>a</sup>)$  activates the urotensin-II receptor (UT), a G-protein-coupled receptor that was originally identified as "orphan" receptor GPR14,<sup>1</sup> and is expressed in many tissues, including blood vessels, heart, liver, kidney, skeletal muscle, and lung.<sup>2</sup> Actually, there is a collection of U-II peptides that range in size from 11 to 14 amino acids across species, from fish to humans, with 11 amino acids in humans to 14 amino acids in mice. Residues 5 through 10 of these U-II peptides comprise a highly conserved CFWKYC cyclic array, whereas the N-terminal region is variable in length and constitution. The WKY motif has been identified as being important for biological activity.<sup>3</sup>

U-II from the urophysis of the goby fish (Gillichthys mirabilis) is a 12-mer cyclic peptide that may well be the most potent vasoconstrictor known.4-<sup>6</sup> This compound caused concentration-dependent contraction of isolated arterial rings of rats and humans with an  $EC_{50}$  less than 1 nM. However, the in vivo effects of U-II can be contradictory because of varied specificity for different tissues and different species. As a case in point, U-II was reported to cause vasodilation in conscious rats.7 Nevertheless, relative to the role of U-II in chronic vascular disease, the goby U-II peptide induced hypertrophy in cardiomyocytes $\frac{8}{3}$  and the proliferation of smooth muscle cells,<sup>9</sup> indicating a potential role in heart

failure and atherosclerosis. From the wealth of knowledge in this area, U-II and its receptor have been implicated in various cardiorenal and metabolic diseases,  $a^{\text{la}-e}$  including hypertension,<sup>10</sup> heart failure,<sup>11</sup> chronic renal failure,<sup>11c,12</sup> diabetes,  $11c,13$  and atherosclerosis.<sup>14</sup> Consequently, we<sup>3b,15</sup> and other researchers<sup>3a,3c,16,17</sup> have been pursuing U-II receptor antagonists as potential therapeutic agents.

To obtain initial lead structures, we devised and executed a high-throughput screening (HTS) protocol involving a functional assay based on cells transfected with rat UT, a fluorometric imaging plate reader (FLIPR) to measure intracellular calcium flux, and the potent, cyclopeptide U-II agonist Ac-CFWK(2-Nal)C-NH<sub>2</sub> (1; rat FLIPR  $EC_{50} = 0.54$  nM, rat UT binding  $K_i = 0.12$  nM) (Chart 1; see Experimental Section).<sup>3b</sup> Application of this assay to a large compound library (∼500000 entities) resulted in several confirmed hits, one of which was moderately potent, nonpeptide antagonist 2 (rat FLIPR IC<sub>50</sub> = 1  $\mu$ M) (Chart 1). Unfortunately, this compound and its more potent bis-desfluoro analogue, 3 (rat FLIPR  $IC_{50} = 150$  nM), turned out to be metabolically unstable to human liver microsomes (HLM  $t_{1/2}$  < 3 min), primarily because of piperazine debenzylation (Chart 1). As part of a drug discovery endeavor, we sought to mitigate this metabolic problem and improve U-II receptor antagonist potency. This effort led to a new class of potent UT antagonists based on piperazino-phthalimide and piperazino-isoindolinone structural subunits (Table 1, compounds  $2-4$ ,  $5a-n$ ,  $6a-j$ ; Table 2, compounds  $7a-o$ ). In this paper, we present our results with these novel series and discuss U-II receptor antagonist pharmacophores.

# Results and Discussion

Synthetic Chemistry. Prototype phthalimide 3 was prepared by the route outlined in Scheme 1. Hydroxyphthalimide I was coupled with veratrylamine in the presence of triethylamine, followed by triflate formation to give II in

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<sup>&</sup>lt;sup>a</sup> Abbreviations: BOP, benzotriazol-1-yloxy-tris(dimethylamino)phosphonium; BSA, bovine serum albumin; CHO, Chinese hamster ovary; dba,  $(E, E)$ -dibenzylideneacetone; FLIPR, fluorometric imaging plate reader; h, human; HBTU, O-benzotriazol-1-yl-N,N,N',N'-tetramethyluronium; HOBt, 1-hydroxybenzotriazole; HLM, human liver microsomes; HTS, high-throughput screening; PK, pharmacokinetics; SAR, structure-activity relationship; TFA, trifluoroacetic acid; Tol-BINAP, 2,2'-bis(di-p-tolyl)phosphino-1,1'-binaphthyl; U-II, urotensin-II; UT, urotensin-II receptor.



excellent yield. Boc-piperazine was introduced by direct displacement in the presence of triethylamine at  $110^{\circ}$ C in a sealed tube in good yield. The Boc group was removed and III was alkylated with benzyl bromide to afford 3 in high yield.

Compounds of structure type 5 were prepared according to the general plan in Scheme 2. Ketones V were obtained from carboxylic acids IV by Weinreb amide formation, followed by addition of 3,4-dimethoxyphenylmagnesium bromide.<sup>18</sup> Reductive amination with ammonium acetate and sodium cyanoborohydride gave primary benzylamines VI, which were converted into phthalimides VII. Activation as a triflate and piperazine displacement yielded target piperazino-phthalimides 5 (Table 1). 2-Thienylsulfonamidoalkyl derivatives 5e, 5g, and 5j were prepared from the corresponding primary amines and 2-thienylsulfonyl chloride with  $K_2CO_3$  as a base. In the case of  $(R, S)$ -5a, an enantioselective synthesis was performed (Scheme 3). Thus, ketone VIII was reduced asymmetrically with borane and prolinol  $IX^{19}$  to alcohol X in high yield, with an optical rotation of  $[\alpha]^{25}$ <sub>D</sub> +42.0 (c 1.07, CHCl<sub>3</sub>). Since Ponzo and Kaufman reported an optical rotation for **X** of  $[\alpha]^{25}$ <sub>D</sub> +40.5  $(c 1.1, CHCl<sub>3</sub>), corresponding to 97% ee<sup>19</sup>, our material was$ considered to be essentially  $\geq$ 99% ee. The mesylate of X was reacted with diallylamine to give XI in good yield with inversion of the stereocenter. The allyl groups were removed by a palladium catalysis procedure<sup>20</sup> and **XII** was transformed into target  $(R, S)$ -5a,  $[\alpha]^{25}$ <sub>D</sub> +13.8 (c 1.18, CH<sub>3</sub>OH), by generation of a phthalimide, formation of a triflate, and displacement with piperazine XIII (8% yield for four steps). On the basis of the chemistry used,  $(R, S)$ -5a should have been >99% ee.

We developed an enantioselective synthesis of key diamine enantiomers  $(R)$ -XVI and  $(S)$ -XVI to provide individual diastereomers for testing (Scheme 4; exemplied for  $(R)$ -XVI). Weinreb amide  $XIV^{16}$  was subjected to asymmetric reductive amination by using ammonium formate and a chiral Binapruthenium complex to give XV (96% ee).<sup>21</sup> The primary amine in  $(R)$ -XV was protected as a trifluoroacetamide, the Cbz group was removed, a Boc group was installed, and the trifluoroacetyl group was removed to afford  $(R)$ -XVI. The incorporation of this intermediate into a final product is illustrated for 7a in Scheme 5. Benzoic acid XVII was esterified and  $\alpha$ -brominated to give XVIII, which was coupled with  $(R)$ -XVI and cyclized thermally to indolinone XIX. The aryl halide was displaced by N-ethylpiperazine under palladium catalysis to give  $XX$ .<sup>22</sup> The Boc group was removed, and the exposed primary amine was coupled with 2-thienylsulfonyl chloride to furnish target compound 7a (96% ee; see Experimental Section).

Development of U-II Receptor Antagonists. As an early approach to metabolic stabilization, we installed an  $\alpha$ -methyl group to generate  $(R)$ -4 (Chart 1). We realized improved metabolic stability (HLM  $t_{1/2} = 29$  min) and improved potency (rat FLIPR  $IC_{50} = 84 \text{ nM}$ );<sup>23</sup> also, (R)-4 proved to be promising in that it exhibited good oral bioavailability in a pharmacokinetics (PK) experiment in rats ( $F=33\%$ ,  $t_{1/2}=$ 4 h). However, in a whole-cell binding assay involving human UT (see Experimental Section),<sup>24</sup> (R)-4 had just moderate affinity (hUT  $K_i$  = 570 nM). Now, the goal became improvement of binding affinity for the human receptor while retaining favorable rat potency (because rats would serve the pharmacological test species) and metabolic stability. We carried out a structure-activity relationship (SAR) study with respect to the 3,4-dimethoxyphenyl segment and found that the 3,4-dimethoxyphenyl group is crucial for decent in vitro potency. Without going into much detail, the following replacements for 3,4-dimethoxyphenyl in prototype  $(R)$ -4 were very unfavorable: phenyl, 3-hydroxy-4-methoxyphenyl, 3-methoxy-4-hydroxyphenyl, 3-ethoxy-4-methoxyphenyl, 3-methoxy-4-ethoxyphenyl, 4-benzoic acid methyl ester, 3-benzoic acid methyl ester, 4-benzoic acid, 3-benzoic acid, 2-pyrimidine, 3-indole, 4-indole, 2-methoxypyridine, and 5,6-dimethoxy-1-indane.

We explored the attachment of substituents onto the methylene portion of the 3,4-dimethoxybenzyl group to probe SAR around that benzylic position. Thus, mono- $\alpha$ substituted 3,4-dimethoxybenzylamines (VI) were converted into phthalimide products 5 (Scheme 2). The rat FLIPR potency of methyl derivative 5a (IC<sub>50</sub> = 70 nM) was in the range of unsubstituted compound 4 (Table 1). Diastereomer  $(R, S)$ -5a had a rat FLIPR IC<sub>50</sub> value of 130 nM, as well as a 3.5-fold potency improvement for the human receptor (hUT  $K_i = 160 \text{ nM}$ ) compared to 4.<sup>25</sup> Ethyl analogue 5b had a rat FLIPR IC<sub>50</sub> value of 160 nM. On the basis of these promising results, we introduced more elaborate substituents at that benzylic position (Table 1). A study of aminoalkyl (5c, 5h, and 5k) and Boc-aminoalkyl (5d, 5f, 5i, and 5l) groups indicated that rat UT antagonist potency was optimal with one to three methylenes in the linker and suggested that human UT potency was optimal with three methylenes in the linker. A follow-up study of 2-thienylsulfonamidoalkyl

Table 1. Urotensin-II Receptor Functional and Binding Data for Phthalimides<sup>a</sup>





 $a$ Target compounds were purified either by silica gel flash-column chromatography (230-400 mesh) or HPLC on a Gilson instrument equipped with a reverse-phase Kromasil C18 column (10  $\mu$ , 250 mm  $\times$  50 mm). All compounds were characterized by ESI-MS and 300 MHz <sup>1</sup>H NMR. Abbreviations: Th, 2-thienyl; CITh, 5-chlorothien-2-yl; Isox, 5-methylisoxazol-4-yl; Imid, 1,2-dimethylimidazol-4-yl; Pyzl, 3,5-dimethylpyrazol-4-yl.<br><sup>b</sup> Inhibition of calcium mobilization in CHO-K1 cells transfected with to UT in cultured human rhabdomyosarcoma cells (RMS13) ( $N = 2$ ).  $\frac{d}{d}$ Inhibition of calcium mobilization in human rhabdomyosarcoma cells, as measured by FLIPR  $(N = 2-6)$ . Corresponds to structure type  $6^{7}$  Reference standard SB-706375 (ref 26; also Dhanak, D.; Gallagher, T. F.; Knight, S. D. PCT Int. Appl. WO 2002089792, 2002). <sup>g</sup> Reference standard palosuran (ref 17i; also Aissaoui, H., et al. PCT Int. Appl. WO 2003048154, 2003).

groups with one to three methylenes (5e, 5g, and 5j) confirmed that a three-carbon linker leads to better human UT antagonist potency (hUT  $K_i = 290$ , 270, and 76 nM, respectively). Methylsulfonamide 5m and, especially, (5-chlorothienyl)sulfonamide 5n showed attenuated potency. Given that such derivatization could yield good antagonist potency against both rat and human receptors, numerous sulfonamides with a three-carbon linker were examined in our rat and human FLIPR-based functional assays and for human UT binding (see Experimental Section). Two notable subunits were  $(5\text{-Me-oxazol-4-yl})SO_2NH(CH_2)_3$  (rat FLIPR IC<sub>50</sub>/hUT K<sub>i</sub>/human FLIPR IC<sub>50</sub> = 14/31/46 nM) and  $MeSO_2CH_2SO_2NH(CH_2)_3-$  (19/25/250 nM), which improved upon 5j (80/76/250 nM).

At this juncture, we pursued the evaluation of single diastereomers only. Although the enantiomers  $(R, S)$ -5j and  $(R, R)$ -5j did not have distinctly different potencies, there was a potency bias for the  $(R, R)$ -isomer over the  $(R, S)$ -isomer (Table 1). Thus, diverse derivatives with the  $(R, R)$  structure were synthesized and tested, and a sampling of our large number of analogues is presented in Table 1. It was found that ethyl (6c) and isopropyl (6d) are optimal groups for the piperazine nitrogen, thereby attaining single-digit nanomolar rat UT antagonist potency. A carboxamide functionality could replace the sulfonamide, albeit with an attenuation in potency (cf. 6e and 6f; 6g and 6h). The piperazine N-substituent and the sulfonamide group did not necessarily combine in an additive manner (cf. 6i and 6j).

## Table 2. Urotensin-II Receptor Functional and Binding Data for Isoindolinones<sup>a</sup>





<sup>a</sup>Target compounds were purified either by silica gel flash-column chromatography (230-400 mesh) or HPLC on a Gilson instrument equipped with a reverse-phase Kromasil C18 column (10  $\mu$ , 250 mm  $\times$  50 mm). All compounds were characterized by ESI-MS and 300 MHz <sup>1</sup>H NMR. Abbreviations: Th, 2-thienyl; ClTh, 5-chlorothien-2-yl; Isox, 5-methylisoxazol-4-yl; Imid, 1,2-dimethylimidazol-4-yl; Pyzl, 3,5-dimethylpyrazol-4-yl. Additional abbreviations: ClPyrm, 6-chloropyrimidin-2-yl; MeOPyrm, 6-methoxypyrimidin-2-yl. <sup>b</sup> Inhibition of calcium mobilization in CHO-K1 cells transfected abbreviations: ClPyrm, 6-chloropyrimidin-2-yl; MeOPyrm, 6-methoxypyrimidin with the rat UT, as measured by FLIPR  $(N = 2-6)$ . <sup>c</sup> Inhibition of  $[^{25}I]U-II$  binding to UT in cultured human rhabdomyosarcoma cells (RMS13)  $(N = 2)$ . <sup>d</sup> Inhibition of calcium mobilization in human rhabdomyosarcoma cells, as measured by FLIPR ( $N = 2-6$ ). <sup>e</sup> Reference standard SB-706375 (ref 26; also Dhanak, D.; Gallagher, T. F.; Knight, S. D. PCT Int. Appl. WO 2002089792, 2002). <sup>f</sup> Reference standard palosuran (ref 17i; also Aissaoui, H., et al. PCT Int. Appl. WO 2003048154, 2003).

#### Scheme 1. Synthesis of 3



We extended our investigation to the corresponding isoindolinone analogues, 7 (Table 2), which were assembled from diamine  $(R)$ -XVI according to the route in Scheme 5. The SAR information gained from the phthalimide series was used as a starting point for the isoindolinone series. Thus, we readily obtained some very potent antagonists, several with single-digit nanomolar potencies in the rat FLIPR assay (viz.  $7a-e$ ,  $7h$ ,  $7j$ ,  $7l$ ,  $7n$ , and  $7o$ ) and some with single-digit nanomolar affinities in the hUT binding assay (viz. 7a, 7b, 7e, 7g, and 7o). Remarkably, 7a also exhibited very potent functional antagonism in the human calcium flux assay ( $IC_{50} = 8.0$  nM) (see Experimental Section). The next tier of compounds in this human cellular functional assay were 7b, 7d, and 7m with  $IC_{50}$  values of 50-60 nM. Across our three assays, ethyl was the optimal N-alkyl group  $(R^2)$  for antagonist potency. It is noteworthy





**Scheme 4.** Synthesis of Intermediate  $(R)$ -XVI



 $R$ -Tol-BINAP =  $(R)$ - $(+)$ -2,2'-bis(di-4-tolylphosphino)-1,1'-binaphthyl

that 7a displayed better potency than two reference U-II receptor antagonists that we tested, 8  $(SB-706375)^{26}$  and palosuran<sup>17i</sup> (9) (Chart 2). In a rat aortic ring assay,  $27$  7a blocked the effect of rat U-II with 74% inhibition at 100 nM, whereas 8 exhibited inhibition of 97% at 100 nM.

Compound 7a was examined in the U-II-induced ear-flush model in rats to determine if it could block a U-II-mediated effect in vivo (Figure 1).<sup>28</sup> Subcutaneous administration of U-II causes an increase in ear pinna temperature, with a maximum effect at 30 min, which can be blocked by

# Scheme 5. Synthesis of 7a



Chart 2. U-II Receptor Antagonist Molecules



antagonist molecules. Compound 7a, administered iv 15 min prior to administration of U-II, attenuated the increase in ear pinna temperature (whereas vehicle had no effect). Complete reversal with 7a in this assay was attained at 10 mg/kg (Figure 1), and a similar response was observed with intravenously administered 8. An area for improvement for 7a and its congeners would be oral bioavailability, as F generally ranged between 10 and 15% in oral PK studies in rats.

To obtain preliminary insight into selectivity for UT over other molecular targets, 7a was profiled for binding affinity in 50 receptors and ion channels at a high concentration of  $10 \mu$ M by CEREP.<sup>29</sup> Less than 50% inhibition of the binding of a reference ligand was observed in 90% of the individual assays.<sup>30</sup> The following assays showed inhibition values of  $>50\%$ :  $\alpha$ 1 adrenergic receptor (68%), human M1 muscarinic receptor (65%), human M3 muscarinic receptor (55%), human 5-HT1A serotonin receptor (93%), 5-HT1B serotonin receptor (66%), and Na<sup>+</sup> channel site 2 (59%).

Different Classes of Nonpeptide U-II Receptor Antagonists. Various U-II receptor antagonist series have been reported.<sup>15-17</sup>

Our earlier series revolved around phenylpiperidine-benzoxazinones, such as 10, which also bears a piperazine group with one basic nitrogen center (Chart 2).15 The series from Actelion was anchored by palosuran (9), in which a 4-quinoline is linked to a 4-substituted piperidine,  $17i$  and GlaxoSmithKline compound 8 contains a 3,4-dimethoxybenzenesulfonamide linked to a pyrrolidine.<sup>26</sup> Some other classes of potent U-II antagonists are represented by structures  $11^{17g}$  and  $12^{17f}$  (Chart 2). There appears to be some analogy between the subunits of chemotypes 10 and 11, but their key basic amine centers do not correlate, as discussed further below.

A cursory inspection of structures  $7a$  and  $8-12$  suggests that there is no clear-cut structural pattern to describe a common pharmacophore for potent UT antagonist ligands. However, to gain a better appreciation of the pharmacophores for these six series, we generated energy-minimized structures for each representative compound (see Supporting Information) with Maestro<sup>31</sup> by employing the OPLS-AA force-field<sup>32</sup> and a GB/SA water model.<sup>33</sup> Each structure was also subjected to a limited amount of conformational analysis by using the mixed-torsion low-mode algorithm<sup>34</sup> in



Figure 1. Effects of 7a on ear pinna temperature over time. A solution of the test compound in 20% aqueous hydroxypropyl- $\beta$ cyclodextrin (HPBDC) was administered intravenously at 3 mg/kg (filled squares) or 10 mg/kg (filled, inverted triangles). The administration of vehicle is depicted by open squares. Data are mean temperature changes from baseline  $\pm$  SEM ( $N = 5$  per group).

Maestro. To assess the relative distances between key interaction sites, a common pharmacophore was generated by using Phase.<sup>35</sup> It was only possible to obtain very simple three-point pharmacophores because of the great diversity between these six structures. The distances in the following discussion were derived from low energy conformations aligned according to this rudimentary pharmacophore. All of the chemotypes possess at least one basic amine nitrogen, which was used as a reference frame, and at least two benzenoid subunits, which were used as secondary reference elements. Compound 7a contains a basic nitrogen center in the piperazine ring that is  $5.5 \text{ Å}$  from the isoindolinone benzene ring (center of ring), 8.3 Å from the amide carbonyl, 9.9 Å from the key dimethoxybenzene ring (center of ring), and  $6.7$  Å from the sulfonamide NH, with the dimethoxybenzene ring (center of ring)  $~6.5$  A from the other benzene ring,  $3.6$  Å from the amide carbonyl, and  $6.4$  Å from the sulfonamide NH. Considering  $8-12$ , they all contain a basic amine nitrogen, as a pyrrolidine, piperidine, piperidine/ piperazine, pyrrolidine, and piperidine/pyrrolidine, respectively, and they all contain at least one aromatic ring that is proximal to the amine center and nearby amide groups. Our series needed the 3,4-dimethoxybenzene subunit for strong antagonist potency, and this subunit is also present in 8; however, it is absent from the series represented by  $9-12$ . The basic center in  $\frac{8}{5.6}$  A from the dimethoxybenzene ring (center of ring), 8.4 A from the sulfonamide NH, and  $6.0 \text{ Å}$ from another aromatic ring. The basic center in  $9$  is 6.4 A from an aromatic ring (center of ring),  $8.9 \text{ Å}$  from the bicyclic aromatic group in the opposite direction, and  $4.5 \text{ Å}$  from the urea carbonyl. There are two basic amine centers in 10, separated by  $\sim$ 10 Å. The benzoxazinone benzene ring is 5.4 Å from the piperidine nitrogen and  $\sim$ 15 Å from the basic piperazine nitrogen; the benzamide benzene ring is  $5.2 \text{ Å}$ from the piperidine nitrogen and  $7.3 \text{ Å}$  from the piperazine amine nitrogen; the piperidine nitrogen is 4.5 A from the benzoxazine carbonyl and 5.7 A from the amide carbonyl; the piperazine amine is  $5.0 \text{ Å}$  from the Boc carbonyl and 4.5 Å from the amide carbonyl. The basic center in 11 is 7.8  $\AA$ from the benzoxazolidinone benzene ring and  $\sim$ 6 Å from the benzenesulfonyl ring; each amide carbonyl is 5.7, 5.8, and 6.7 Å from the amine center. The pyrrolidine nitrogen in 12 is 7.3 A from the benzamide ring and 3.8 A from the benzyl ring; the piperidine nitrogen is 6.6 A from the benzyl ring and



Figure 2. Spatial arrangement and distances for loci in the groundstate conformation of tricycle 13. The basic piperazine nitrogen (N), carbonyl oxygen (O), and center of the isoindolinone benzene ring (A) are coplanar, while the center of the dimethoxybenzene (B) projects forward from that defined plane.

 $\sim$ 17 A from the benzamide ring; the amide carbonyl is 4.6 A from the pyrrolidine nitrogen and  $\sim$ 14.5 A from the piperidine nitrogen. This comparative analysis, albeit less than rigorous, suggests that several pharmacophore models can satisfy the requirements for U-II receptor binding and antagonism. A caveat to this analysis would be that we have only considered unbound, ground-state conformations (global energy minima), whereas receptor-bound conformations may be different and more relevant. Given this assessment, it would appear that effective UT antagonists are likely to emanate from diverse chemical entities with specific pharmacophore components related to those that are highlighted.

A key issue in dealing with ligands  $7a$  and  $8-12$  to establish understanding about UT pharmacophore models is the conformational freedom available to these molecules. Their wide structural variation also poses a problem. It would be useful to consider a UT ligand with much less conformational dimensionality. Fortunately, we identified tricyclic compound 13 as a potent UT antagonist (rat FLIPR IC<sub>50</sub> = 14 nM; human UT binding  $K_i$  = 64 nM), which is quite conformationally constrained (Chart 2). $36$  In its groundstate conformation, 13 has its basic amine nitrogen 5.5  $\AA$ from the isoindolinone benzene ring (center of ring),  $\sim$ 9 Å from the amide carbonyl, and  $\sim$ 10 A from the dimethoxybenzene ring (center of ring), with the dimethoxybenzene ring spaced  $4.5 \text{ Å}$  from the amide carbonyl and the two benzene rings (centers) separated by  $6.5 \text{ Å}$  (Figure 2). Many of these dimensions correspond to the dimensions in 7a (disregarding the sulfonamide group).

On the basis of antagonist structures known through 2005, Carotenuto et al. described two general pharmacophores in terms of a basic amine linked to an aryl-1 and to an aryl-2 group, or an aryl-1 linked to an aryl-2 that is then linked to a basic amine.<sup>16b</sup> Lescot et al. also developed more than one antagonist pharmacophore, which suggests a lack of convergence between the antagonist ligands that were considered.<sup>37</sup> They advanced two different models comprised of two aromatic rings, one hydrophobic group, a basic amine center, and a hydrogen-bond acceptor, each with a different spatial arrangement.<sup>37</sup> Consequently, there would seem to be general agreement that UT antagonist structures can be achieved through more than one pharmacophore model. Given that 13 has a fairly well-defined spatial geometry, because of its conformational constraints,<sup>36</sup> it serves to define one type of UT antagonist pharmacophore.

## Conclusion

We have discovered two related series of nonpeptide U-II receptor antagonists based on piperazino-phthalimide (5 and 6) and piperazino-isoindolinone (7) scaffolds. Compound 7a was found to be an exceedingly potent U-II receptor antagonist in vitro, and it also blocked the effects of U-II in vivo in the rat ear-flush model. Structure types 6 and 7 are very distinctive relative to the structure types for other nonpeptide U-II receptor antagonist series, such as those represented by  $8^{26}$  and palosuran (9).<sup>17i</sup> Our series required the 3,4-dimethoxybenzene subunit for robust antagonist potency, and this structural element is also present in 8. However, the series represented by  $9-12$  do not possess this subunit. In any case, potent UT antagonists contain at least one basic amine nitrogen and at least two benzenoid subunits, with one aromatic ring being proximal to the amine center, as well as amide groups near the amine center. Interestingly, such structural elements are contained within the U-II agonist pharmacophore, although with much different distances. $3$  At this juncture, we can conclude that several pharmacophore models will satisfy the requirements for U-II receptor binding and antagonist action.

# Experimental Section

General Chemical Procedures. <sup>1</sup>H NMR spectra were generally acquired at 300.14 MHz on a Bruker Avance-300 spectrometer in  $CDCl<sub>3</sub>$  with Me<sub>4</sub>Si as an internal standard, unless indicated otherwise. The 400-MHz <sup>1</sup>H NMR spectra were recorded with a Bruker AC-400 spectrometer with Me4Si as an internal standard. Electrospray (ES) mass spectra were obtained on a Micromass Platform LC single quadrupole mass spectrometer in the positive mode. The structures of all new compounds were consistent with their <sup>1</sup>H NMR and ES-MS mass spectra. Analytical HPLC analyses were performed on an Agilent series 1100 HPLC instrument eluting with a gradient of water/MeCN/TFA (10:90:0.2 to 90:10:0.2) over 4 min with a flow rate of 0.75 mL/min on a Kromasil C18 column (50 mm  $\times$ 2.0 mm; 3.5  $\mu$ m particle size) or on a Supelcosil ABZ+Plus column (50 mm  $\times$  2.0 mm; 3.5  $\mu$ m particle size) at 32 °C, with signals being recorded simultaneously at 220 and 254 nm with a diode-array detector. These HPLC analyses served as the principal determinant of compound purity, which was >95% for target compounds. The measured purity for target compounds is given in parentheses throughout the Experimental Section. Normal-phase preparative chromatography was performed on an Isco Combiflash Separation System Sg 100c equipped with a Biotage FLASH Si 40 M silica gel cartridge (KP-Sil Silica, 32-  $63 \mu$ ,  $60 \text{ Å}$ ; 4 cm  $\times$  15 cm) eluting at 35 mL/min with detection at 254 nm. Reversed-phase preparative chromatography was performed on a Gilson HPLC with a reversed-phase Kromasil column (10  $\mu$ , 100 Å C18, column length 250 mm  $\times$  50 mm). Optical rotations were measured on a Perkin-Elmer 241 polarimeter. Elemental analysis and Karl Fischer water analysis were determined for certain specific compounds by Quantitative Technologies Inc., Whitehouse, NJ.

Synthesis of 4-Hydroxy-2-(3,4-dimethoxybenzyl)-1,3-dioxo-2,3 dihydro-1H-isoindol-4-yl Trifluoromethanesulfonic Acid Ester (II). A 200 mL round-bottom flask was charged with 4-hydroxy-2,3 dihydro-1H-isobenzofuran-1,3-dione  $(3.07 \text{ g}, 0.019 \text{ mol})$  and dry toluene (94 mL). Veratrylamine (2.8 mL, 0.019 mol) and triethylamine (3.6 mL, 0.026 mol) were added to the mixture. A Dean-Stark trap filled with 4A molecular sieves was attached to the flask and the mixture was refluxed for 24 h. The mixture was cooled to room temperature and diluted with  $CH_2Cl_2$  (200 mL) and washed with 1 N HCl (100 mL). The organic layers were combined and dried using MgSO4, filtered through celite, and concentrated in vacuo to give 5.6 g (96%) of a yellow solid. <sup>1</sup>H NMR  $\delta$  7.70–7.86 (m, 1 H), 7.57 (overlapping dd, J=7.5 Hz, 1 H), 7.37 (d, J=7.3 Hz, 1 H), 7.15 (d,  $J = 8.4$  Hz, 1 H), 6.91 - 7.01 (m, 2 H), 6.81 (d,  $J =$ 7.8 Hz, 1 H), 4.74 (s, 2 H), 3.88 (s, 3 H), 3.85 (s, 3 H). MS ( $ES^+$ )  $314 (M + 1)$ . HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ),

gradient  $10-90\%$  (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R = 3.38$  min (98%). A 500 mL round-bottom flask containing 4-hydroxy-2-(3,4-dimethoxybenzyl)-2,3-dihydro-1Hisoindole-1,3-dione (5.90 g, 0.019 mol),  $CH_2Cl_2$  (80 mL), and  $Et<sub>3</sub>N$  (3.6 mL, 0.026 mol) was cooled using an ice-water bath. A solution of trifluoromethanesulfonyl chloride (2.2 mL, 0.021 mol) in  $CH_2Cl_2$  (20 mL) was added dropwise via an addition funnel. The mixture was then stirred for 1 h in an ice-water bath. The mixture was then diluted with  $CH_2Cl_2 (200 \text{ mL})$  and washed with 1 N HCl (100 mL). The organic layer was dried using MgSO4, filtered through celite, and concentrated in vacuo to give 8.20 g (97%) of **II** as a yellow solid. <sup>1</sup>H NMR  $\delta$  7.78–7.87 (m, 2 H), 7.53 (d,  $J = 8.3$  Hz, 1 H),  $6.99 - 7.02$  (m, 2 H),  $6.79$  (d,  $J =$ 8.1 Hz, 1 H), 4.79 (s, 2 H), 3.87 (s, 3 H), 3.84 (s, 3 H). MS ( $ES^+$ ) 446 (M + 1). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient 10-90% (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R = 4.14$  min (96%).

General Synthesis of Compounds 2 and 3. 4-[4-(Benzyl) piperazin-1-yl]-2-(3,4-dimethoxybenzyl)-2,3-dihydro-1H-isoindole-1,3-dione (3). A 10 mL sealed-tube reactor was charged with II (0.50 g, 1.12 mmol), 1-Boc-piperazine (0.22 g, 1.18 mmol), toluene  $(1.3 \text{ mL})$ , and Et<sub>3</sub>N  $(0.2 \text{ mL}, 1.43 \text{ mmol})$ . The tube was sealed under argon and heated to 110  $\degree$ C for 18 h. The mixture was cooled to 23 °C and purified via flash silica gel chromatography (230-400 mesh silica gel 60, 80:20 hexanes:ethyl acetate) to give 0.33 g (61%) of a yellow solid. <sup>1</sup>H NMR (300 MHz, MeOH)  $\delta$  7.67 (overlapping dd, J=7.2 Hz, 1 H), 7.38 (d, J=7.2 Hz, 1 H), 7.28 (d,  $J=8.2$  Hz, 1 H), 6.99 (s, 1 H), 6.86–6.93 (m, 2 H), 4.71 (s, 2 H), 3.81 (s, 3 H), 3.79 (s, 3 H), 3.61-3.65 (m, 4 H), 3.27-3.31 (m, 4 H), 1.48 (s, 9 H). MS ( $ES^+$ ) 482 (M + 1). A 50 mL round-bottom flask containing 4-[2-(3,4-dimethoxybenzyl)-1,3-dioxo-2,3-dihydro-1H-isoindol-4-yl]-piperazine-1-carboxylic acid tert-butyl ester  $(0.33 \text{ g}, 0.69 \text{ mmol})$  and CH<sub>2</sub>Cl<sub>2</sub>  $(3.0 \text{ mL})$  was treated with TFA (0.7 mL) and the mixture was stirred at 23  $^{\circ}$ C for 1 h. The mixture was concentrated in vacuo, and the crude oil was dissolved in THF (3.4 mL). Benzyl bromide (0.12 mL, 0.97 mmol) and diethylisopropylamine (0.26 mL, 1.49 mmol) were added. The mixture was heated at reflux for 24 h and evaporated. The crude oil was purified on a Gilson HPLC with a reversed phase Kromasil C18 column (gradient  $80:20-0:100 \text{ H}_2\text{O}$ :MeCN) to give 369 mg (85%) of 3 as a yellow solid. <sup>1</sup>H NMR  $\delta$  7.60 (overlapping dd,  $J=$ 7.4 Hz, 1 H),  $7.39 - 7.50$  (m, 6 H),  $7.12$  (d,  $J = 8.2$  Hz, 1 H),  $6.90 -$ 6.97 (m, 2 H), 6.37 (d, J=8.7 Hz, 1 H), 4.72 (s, 2 H), 4.27 (s, 2 H), 3.87 (s, 3 H), 3.83 (s, 3 H), 3.66-3.77 (m, 4 H), 3.41-3.48 (m, 2 H), 3.14-3.18 (m, 2 H). MS ( $ES^{+}$ ) 472 (M + 1). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient 10-90% (MeCN + 0.16% TFA) in (water  $+$  0.2% TFA) over 4 min,  $t_R$  = 3.21 min  $(99\%)$ .

General Synthesis of Compounds 4 and 5b-n. N-(2-Thiophenesulfonyl)-[4-(3,4-dimethoxyphenyl)]-4-{1,3-dioxo-4-[4-(1 phenylethyl)piperazin-1-yl]-2,3-dihydro-1H-isoindol-2-yl}butyl) amide) (5j). A 500 mL round-bottom flask containing DMF (207 mL) and 4-tert-butoxycarbonylaminobutyric acid (4.2 g, 20.6 mmol) was cooled using an ice-water bath. Triethylamine (8.8 mL, 63.1 mmol) was added, followed by the benzotriazol-1-yloxy-tris(dimethylamino)phosphonium (BOP) hexafluorophosphate (10.0 g, 22.6 mmol) and N,O-dimethylhydroxylamine hydrochloride (3.1 g, 31.8 mmol). The mixture was stirred at room temperature for 24 h, diluted with  $CH_2Cl_2$  (500 mL), and transferred to a separatory funnel. The organic layer was washed with 1 N HCl (300 mL) and water  $(2 \times 300 \text{ mL})$ . The organic layer was dried using MgSO4, filtered through celite, concentrated in vacuo, and purified via flash chromatography (230-400 mesh silica gel 60, gradient  $100:0-90:10 \text{ CH}_2\text{Cl}_2$ :MeOH) to give 5.08 g of [3-(methoxymethylcarbamoyl)propyl]carbamic acid tertbutyl ester as a white solid (91%). <sup>1</sup>H NMR  $\delta$  3.65 (s, 3 H), 3.22-3.21 (m, 2 H), 3.16 (s, 3 H), 2.45-2.54 (m, 2 H), 1.80-1.89 (m, 2 H), 1.42 (s, 9 H). HPLC: Kromasil C18 column  $(50 \text{ mm} \times 2 \text{ mm}; 3.5 \mu)$ , gradient  $10-90\%$  (MeCN  $+0.16\%$  TFA) in (water  $+0.2\%$  TFA) over 4 min,  $t_R = 2.74$  min (97%). A 1 L round-bottom flask

containing this material (5.09 g, 20.7 mmol) and tetrahydrofuran (415 mL) was cooled by using an ice-water bath. A solution of 3,4 dimethoxyphenylmagnesium bromide in THF (207 mL, 104 mmol) was added dropwise via an addition funnel over 30 min. The mixture was allowed to warm to room temperature and stirred for 20 h. Water (150 mL) was added, and the mixture was concentrated in vacuo, extracted with  $CH_2Cl_2$  (600 mL), and washed with water  $(2 \times 300 \text{ mL})$ . The organic layer was dried with MgSO4, filtered through celite, concentrated in vacuo, and purified via flash chromatography (230-400 mesh silica gel 60, gradient 90:10-50:50 hexanes:ethyl acetate) to give 4.0 g of [4-(3,4 dimethoxyphenyl)-4-oxo-butyl]carbamic acid tert-butyl ester [V,  $R<sup>1</sup> = (CH<sub>2</sub>)<sub>3</sub>NH-Boc]$ , as a white solid (60%). <sup>1</sup>H NMR  $\delta$  7.57 (d,  $J=8.2$  Hz, 1 H),  $7.52$  (s, 1 H),  $6.88$  (d,  $J=8.6$  Hz, 1 H),  $3.95$  (s, 3 H), 3.92 (s, 3 H), 3.27-3.34 (m, 2 H), 2.97-3.02 (m, 2 H), 1.92-1.99  $(m, 2 H)$ , 1.42 (s, 9 H). HPLC: Kromasil C18 column (50 mm  $\times$  $2 \text{ mm}$ ;  $3.5 \mu$ ), gradient  $10 - 90\%$  (MeCN  $+ 0.16\%$  TFA) in (water  $+$ 0.2% TFA) over 4 min,  $t_R = 3.48$  min (98%). A 300 mL roundbottom flask was charged with this material (4.0 g, 12.4 mmol), ammonium acetate (9.5 g, 123.2 mmol), and methanol (42.0 mL). Sodium cyanoborohydride (0.55 g, 8.8 mmol) was added, and the mixture was heated at 40  $\degree$ C for 22 h. The mixture was cooled to room temperature, and 1 N NaOH (200 mL) was added. The mixture was transferred to a separatory funnel and extracted with  $CH_2Cl_2$  (3  $\times$  200 mL). The organic layer was dried with MgSO<sub>4</sub>, filtered through celite, concentrated in vacuo, and purified via flash chromatography (230-400 mesh silica gel 60, gradient 100:0-90:10  $CH_2Cl_2$ :MeOH) to give 2.0 g of [4-amino-4-(3,4-dimethoxyphenyl)butyl]carbamic acid *tert*-butyl ester [VI,  $R^1 = (CH_2)_3NH- Boc$ ], as a white solid (50%). <sup>1</sup>H NMR  $\delta$  6.83–6.87 (m, 3 H), 4.53–4.60  $(m, 1 H), 3.90 (s, 3 H), 3.87 (s, 3 H), 3.10-3.15 (m, 2 H), 1.63-1.71$  $(m, 5 H)$ , 1.43 (s, 9 H). HPLC: Kromasil C18 column (50 mm  $\times$  $2 \text{ mm}$ ;  $3.5 \mu$ ), gradient  $10 - 90\%$  (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R = 2.84$  min (97%). To a 50 mL roundbottom flask containing 4-hydroxy-2,3-dihydro-1H-isobenzofuran-1,3-dione (1.1 g, 6.70 mmol) and dry toluene (30 mL) was added [4-amino-4-(3,4-dimethoxyphenyl)butyl]carbamic acid tert-butyl ester (2.2 g, 6.79 mmol) and triethylamine (1.2 mL, 8.61 mmol). A Dean-Stark trap filled with 4A molecular sieves was attached to the flask and the mixture was refluxed for 24 h. The mixture was cooled to room temperature, diluted with  $CH_2Cl_2$  (200 mL), and washed with 1 N HCl (100 mL). The organic layer was dried (MgSO4), filtered through celite, and concentrated in vacuo to give 3.1 g (98%) of [4-(3,4-dimethoxyphenyl)-4-(4-hydroxy-1,3-dioxo-2,3-dihydro-1H-isoindol-2-yl)butyl]carbamic acid tert-butyl ester [VII,  $R^1 = (CH_2)_3NH$ -Boc], as a yellow solid. <sup>1</sup>H NMR  $\delta$  7.66–7.70 (m, 1 H), 7.52-7.57 (m, 1 H), 7.06-7.16 (m, 1 H), 6.79-6.88 (m, 3 H), 5.16-5.21 (m, 1 H), 3.88 (s, 3 H), 3.85 (s, 3 H), 3.06-3.19 (m, 1 H), 2.49-2.56 (m, 1 H), 2.26-2.33 (m, 1 H), 1.46-1.56 (m, 2 H), 1.42 (s, 9 H). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient  $10-90\%$  (MeCN  $+0.16\%$  TFA) in (water  $+0.2\%$  TFA) over 4 min,  $t_R = 3.94$  min (97%). A 500 mL round-bottom flask containing this material (6.60 g, 0.014 mol),  $CH_2Cl_2$  (60 mL), and  $Et<sub>3</sub>N$  (2.3 mL, 0.017 mol) was cooled using an ice-water bath. A solution of trifluoromethanesulfonyl chloride (1.6 mL, 0.015 mol) in  $CH<sub>2</sub>Cl<sub>2</sub>$  (10 mL) was added dropwise via an addition funnel. The mixture was stirred for 1 h in an ice-water bath. The mixture was diluted with  $CH_2Cl_2(200 \text{ mL})$ , washed with 1 N HCl (100 mL), and saturated aqueous sodium bicarbonate  $(1 \times 100 \text{ mL})$ . The organic layer was dried using MgSO<sub>4</sub>, filtered through celite, and concentrated in vacuo to give 7.06 g  $(84%)$  of triflate as a yellow solid. <sup>1</sup>H NMR  $\delta$  7.76-7.83 (m, 2 H), 7.51 (d, J = 7.8 Hz, 1 H), 7.07-7.12 (m, 2 H), 6.80 (d, J=8.0 Hz, 1 H), 5.22-5.27 (m, 1 H), 3.88 (s, 3 H), 3.85  $(s, 3 H), 3.16-3.26$  (m, 2 H), 2.42-2.59 (m, 1 H), 2.29-2.40 (m, 1 H), 1.55-1.66 (m, 2 H), 1.42 (s, 9 H). MS (ES<sup>+</sup>) 603 (M + 1). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient  $10-90\%$  (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R = 4.72$  min (96%). A 50 mL sealed-tube vessel was charged with this material  $(7.53 \text{ g}, 0.013 \text{ mol})$ , 1- $[(R)-1$ -phenylethyl]piperazine (2.50 g, 0.013 mol), toluene (13 mL), and  $Et_3N(2.4)$ 

mL, 0.017 mol). The tube was sealed under argon and heated to 110 C for 21 h. The mixture was cooled to room temperature and purified via flash silica gel chromatography (230-400 mesh silica gel 60, 80:20 hexanes: ethyl acetate) to give 4.61 g  $(57%)$  of  $(4-(3, 47))$ 4-dimethoxyphenyl)-4-{1,3-dioxo-4-[4-(1-phenylethyl)piperazin-1 yl]-2,3-dihydro-1*H*-isoindol-2-yl}butyl)carbamic acid *tert*-butyl ester (5i) as a yellow solid. <sup>1</sup>H NMR  $\delta$  7.48–7.54 (m, 1 H), 7.23–7.36  $(m, 7 H), 7.07 - 7.11$   $(m, 2 H), 6.78$   $(d, J = 8.2 Hz, 1 H), 5.18 - 5.23$  $(m, 1 H), 4.08 - 4.16$   $(m, 1 H), 3.87$  (s, 3 H), 3.83 (s, 3 H), 3.26 - 3.45 (m, 4 H), 3.14-3.22 (m, 2 H), 2.72-2.76 (m, 2 H), 2.55-2.64 (m, 2 H), 2.44-2.52 (m, 1 H), 2.23-2.30 (m, 1 H), 1.38-1.54 (m, 14 H).  $MS (ES<sup>+</sup>) 643 (M + 1)$ . HPLC: Kromasil C18 column (50 mm  $\times$  $2 \text{ mm}$ ; 3.5  $\mu$ ), gradient 10-90% (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R = 3.31$  min (97%). A 50 mL roundbottom flask was charged with  $5i$  (1.90 g, 2.95 mmol) and 1,4dioxane (15 mL). A solution of 4 N HCl in 1,4-dioxane (10 mL) was added, and the mixture was stirred at room temperature for 4 h. The mixture was concentrated in vacuo to give 5h. <sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD)  $\delta$  7.63-7.68 (m, 1 H), 7.51-7.61 (m, 5 H), 7.41 (d, J = 7.2 Hz, 1 H), 7.28 (d,  $J = 7.9$  Hz, 1 H), 7.13 (s, 1 H), 7.04 (d,  $J =$ 6.5 Hz, 1 H), 6.89 (d,  $J=8.4$  Hz, 1 H), 5.22-5.27 (m, 1 H), 4.46-4.57 (m, 1 H), 3.82 (s, 3 H), 3.80 (s, 3 H), 3.47-3.66 (m, 4 H), 3.13- 3.26 (m, 2 H), 2.96-3.07 (m, 2 H), 2.57-2.67 (m, 1 H), 2.32-2.42  $(m, 1 H)$ , 1.80  $(d, J=6.8 Hz, 1 H)$ , 1.64-1.74  $(m, 2 H)$ , 1.29-1.35  $(m, 2 H)$ . MS  $(ES^+)$  543  $(M + 1)$ . HPLC: Kromasil C18 column  $(50 \text{ mm} \times 2 \text{ mm}; 3.5 \mu)$ , gradient  $10-90\%$  (MeCN + 0.16% TFA) in (water  $+ 0.2\%$  TFA) over 4 min,  $t_R = 2.44$  min (98%). The solid  $(1.0 \text{ g}, 1.63 \text{ mmol})$  was dissolved in  $\text{CH}_2\text{Cl}_2$  (8.2 mL). Triethylamine (0.7 mL, 5.02 mmol) was added followed by 2-thiophenesulfonyl chloride (0.37 g, 2.03 mmol). The mixture was stirred at room temperature for 24 h and concentrated in vacuo. The crude oil was purified on a Gilson HPLC with a reverse-phase Kromasil C18 column (gradient 80:20-0:100  $H<sub>2</sub>O$ :MeCN). The purified material was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (50 mL), treated with 1 N HCl in diethyl ether (10 mL), and concentrated in vacuo. This procedure was repeated two more times to give 1.18 g (96%) of 5j as a yellow solid. <sup>1</sup>H NMR  $\delta$ 7.71-7.74 (m, 1 H), 7.51-7.61 (m, 3 H), 7.29-7.38 (m, 6 H), 7.02-7.18 (m, 3 H), 6.73-6.80 (m, 1 H), 5.10-5.17 (m, 1 H), 4.50-4.57 (m, 1 H), 3.86 (s, 3 H), 3.83 (s, 3 H), 3.31-3.39 (m, 4 H), 3.06-3.11 (m, 4 H), 2.65-2.74 (m, 2 H), 2.45-2.55 (m,  $1 \text{ H}$ ), 2.20-2.31 (m, 1 H), 1.58-1.63 (m, 2 H), 1.49 (d,  $J=6.8 \text{ Hz}$ , 3 H). MS ( $ES^+$ ) 689 (M + 1). HPLC: Kromasil C18 column  $(50 \text{ mm} \times 2 \text{ mm}; 3.5 \mu)$ , gradient  $10-90\%$  (MeCN  $+ 0.16\%$ TFA) in (water  $+ 0.2\%$  TFA) over 4 min,  $t_R = 2.54$  min (100%).

Synthesis of  $(R, S)$ -5a and  $(R, R)$ -5a. A 50 mL round-bottom flask was charged with borane-dimethyl sulfide solution in toluene (0.4 mL, 0.8 mmol) and toluene (6.0 mL). Octanol (0.32 mL, 2.01 mmol) was added dropwise to the mixture, and stirring was continued at 34 °C for 1 h.  $(S)$ - $\alpha$ , $\alpha$ -Diphenyl-2pyrrolidinemethanol (0.14 g, 0.55 mmol) was dissolved in toluene (6.0 mL) and added to the mixture, followed by stirring for 1 h at  $34^{\circ}$ C. Borane-dimethyl sulfide solution in toluene (2.8) mL, 1.40 mmol) was added, followed by dropwise addition of a solution of 3,4-dimethoxyacetophenone (1.0 g, 5.55 mmol) in toluene (6.0 mL) over 1 h via addition funnel. The mixture was cooled to room temperature and quenched with 1 N HCl (20 mL). The mixture was transferred to a separatory funnel and extracted with ethyl acetate (200 mL). The organic layer was washed with sodium bicarbonate (50 mL) and sodium chloride (50 mL). The organic layer was dried using  $MgSO<sub>4</sub>$ , filtered through celite, and concentrated in vacuo to give 820 mg of  $(R)$ -1-(3,4-dimethoxyphenyl)ethanol (X) as a white solid. <sup>1</sup>H NMR  $\delta$  6.82–6.95 (m, 3 H), 4.86 (q, J = 3.3 Hz, 2 H), 3.90 (s, 3 H), 3.88 (s, 3 H), 1.83 (s, 1 H), 1.49 (d,  $J = 6.4$  Hz, 3 H);  $[\alpha]^{25}D + 42.0$  $(c \ 1.07, \ CHCl<sub>3</sub>)$ . A 100 mL round-bottom flask containing  $X$  (0.40 g, 2.20 mmol) and  $CH_2Cl_2$  (11.0 mL) was cooled using an ice-water bath. Triethylamine (0.37 mL, 2.65 mmol) was added to the mixture, followed by the dropwise addition methanesulfonyl chloride (0.19 mL, 2.45 mmol). The mixture was stirred for 4 h at  $0^{\circ}$ C and then washed with 1 N HCl (5 mL). The organic layer was dried with MgSO<sub>4</sub> and filtered through celite. Diallylamine (0.27 mL, 2.19 mmol) and 2,2,6,6-tetramethylpiperidine (0.82 mL, 4.83 mmol) were added to the crude  $CH_2Cl_2$ solution. The mixture was refluxed for 24 h, cooled to room temperature, and concentrated in vacuo. The crude oil was purified via flash chromatography (230-400 mesh silica gel 60, gradient  $100:0-98:2 \text{ CH}_2\text{Cl}_2$ :MeOH) to give 0.4 g of diallyl- $[1-(R)-(3,4-dimethoxyphenyl)ethyl]$ amine (XI) as a white solid. <sup>1</sup>H NMR  $\delta$  6.78-6.87 (m, 3 H), 5.78-5.91 (m, 2 H),  $5.09 - 5.20$  (m, 4 H), 4.18 (q,  $J = 6.4$  Hz, 2 H), 3.90 (m, 3 H), 3.86  $(m, 3 H), 3.00 - 3.17$   $(m, 4 H), 1.45$   $(d, J = 6.4 Hz, 3 H).$  A 50 mL round-bottom flask was charged with XI (0.40 g, 1.53 mmol) and  $CH_2Cl_2$  (8.0 mL).  $Pd_2(dba)_3 \cdot CHCl_3$  (0.16 g, 0.15 mmol), triphenylphosphine (0.16 g, 6.1 mmol), and 1,3-dimethylbarbituric acid (0.79 g, 5.06 mmol) were added, and the mixture was heated at reflux for 3 h. The mixture was cooled to room temperature, transferred to a separatory funnel, and extracted with 1 N HCl (50 mL). The aqueous layer was basified with 3 N NaOH and extracted with ethyl acetate (200 mL). The organic layer was dried with  $MgSO_4$  and filtered through celite to give  $1-(R)-(3,4\t{-}dimethoxyphenyl)ethylamine (XII)$  as a white solid. <sup>1</sup>H NMR  $\delta$  6.74–6.85 (m, 3 H), 3.96–4.08 (m, 1 H), 3.83 (m, 3 H), 3.80 (m, 3 H), 1.87-1.97 (broad s, 2 H), 1.31 (d, J=6.6 Hz, 3 H). Compound  $(R, S)$ -5a was prepared by the general methods for compounds  $4$  and  $5b-n$  (vide supra) but employing enantiomerically pure benzylamine  $XII(0.14 g)$ . Compound  $(R, S)$ -5a was isolated as a yellow solid. <sup>1</sup>H NMR  $\delta$  7.35–7.54 (m, 7 H),  $6.93 - 7.07$  (m, 3 H),  $6.73$  (d,  $J = 8.6$  Hz, 1 H),  $5.36$  (q,  $J = 7.2$  Hz, 1 H), 4.21 (q, J=5.2 Hz, 1 H), 3.79 (s, 3 H), 3.77 (s, 3 H), 3.36-3.64  $(m, 6 H), 3.79 - 3.16$   $(m, 2 H), 1.83$   $(d, J = 7.0$  Hz, 3 H $), 1.79$   $(d, J = 1.79)$ 7.4 Hz, 3 H). MS (ES<sup>+</sup>) 500 (M + 1);  $[\alpha]^{25}$  p +13.8 (c 1.18, CH<sub>3</sub>OH). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient  $10-90\%$  (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R = 3.43$  min (97%).

Synthesis of  $[(R)-4-Amino-4-(3,4-dimethoxyphenyl)butyl] car$ bamic Acid tert-Butyl Ester (XVI). A 1 L round-bottom flask containing [3-(methoxymethylcarbamoyl)propyl]carbamic acid benzyl ester (5.08 g, 20.7 mmol) and tetrahydrofuran (415 mL) was cooled using an ice-water bath. A solution of 3,4-dimethoxyphenylmagnesium bromide in THF (207 mL, 104 mmol) was added dropwise via addition funnel over 30 min. The mixture was allowed to warm to room temperature and stirred for 20 h. Water (150 mL) was added to the mixture and concentrated in vacuo. The mixture was extracted with  $CH_2Cl_2$  $(600 \text{ mL})$  and washed with water  $(2 \times 300 \text{ mL})$ . The organic layer was dried with  $MgSO<sub>4</sub>$ , filtered through celite, concentrated in vacuo, and purified via flash chromatography (230- 400 mesh silica gel 60, gradient 90:10-50:50 hexanes:ethyl acetate) to give 4.0 g of [4-(3,4-dimethoxyphenyl)-4-oxo-butyl]carbamic acid benzyl ester as a white solid. <sup>1</sup>H NMR  $\delta$  7.57 (d, J = 8.4 Hz, 1 H), 7.51 (s, 1 H), 7.29–7.44 (m, 5 H), 6.88 (d,  $J=8.4$  Hz, 1 H), 5.08 (s, 2 H), 3.95 (s, 3 H), 3.93 (s, 3 H), 3.27-3.34 (m, 2 H), 2.97-3.02 (m, 2 H), 1.92-2.01 (m, 2 H). MS  $(ES^+)$  358 (M + 1). HPLC: Kromasil C18 column (50 mm  $\times$ 2 mm; 3.5  $\mu$ ), gradient 10-90% (MeCN + 0.16% TFA) in (water  $+$  0.2% TFA) over 4 min,  $t_R$  = 3.62 min (98%). A 200 mL Schlenk tube was charged with  $[RuCl_2(benzene)]_2$  (2.0 g, 4.0 mmol) and  $(R)$ -Tol-BINAP  $(5.7 \text{ g}, 8.4 \text{ mmol})$ . The tube was put under vacuum for 15 min and then backflushed with argon. DMF (133 mL, degassed with argon) was added to the tube, and the mixture was flushed with argon. The tube was sealed off and heated to 100  $^{\circ}$ C for 10 min (with stirring). The DMF was then removed under high vacuum at  $70^{\circ}$ C to give  $[(R)-T_0I-BINAP)RuCl_2(DMF)_x]$  as a reddish/brown solid.<sup>3</sup> 200 mL sealed tube was charged with [4-(3,4-dimethoxyphenyl)- 4-oxo-butyl]carbamic acid benzyl ester (8.66 g, 24.2 mmol),  $[(R)-T_0I-BINAP)RuCl<sub>2</sub>(DMF)<sub>x</sub>](2.1 g, 2.5 mmol)$ , ammonium formate (15.3 g, 243 mmol), and a 2.0 M solution of ammonia in methanol (97 mL). The tube was flushed with argon, sealed, and

heated at 85  $\degree$ C for 22 h. The mixture was cooled to room temperature, and the sealed tube was opened carefully due to the release of pressure from excess ammonia. The mixture was concentrated in vacuo, diluted with 1 N HCl (300 mL) and ethanol (150 mL), heated at reflux for 2 h, cooled to room temperature, and washed with diethyl ether  $(1 \times 500 \text{ mL})$ . The aqueous layer was basified with 3 N NaOH to pH  $> 10$ . It was extracted by using  $CH_2Cl_2$  (3  $\times$  400 mL), and the organic solution was dried  $(MgSO<sub>4</sub>)$ , filtered through celite, and concentrated in vacuo to give 15.66 g of  $[4-(R)$ -amino-4- $(3,4$ dimethoxyphenyl)butyl]carbamic acid benzyl ester XV as a white solid. <sup>1</sup>H NMR  $\delta$  7.30–7.34 (m, 5 H), 6.81–6.87 (m, 3 H), 5.08 (s, 2 H), 3.83-3.90 (m, 7 H), 3.16-3.22 (m, 2 H), 1.38-1.71 (m, 6 H). MS ( $ES^{+}$ ) 359 (M + 1). Daicel Chiralpak AD-H,  $4.6 \text{ mm} \times 15 \text{ cm}$ , hexanes: 2-propanol:  $Et_2NH(86:14:0.1)$ 1.0 mL/min; (S)-enantiomer 13.57 min, (R)-enantiomer 15.67 min; 96% ee. A 500 mL round-bottom flask containing  $XV$  (15.66 g, 0.044 mol) and  $CH_2Cl_2$  (220 mL) was cooled using an ice-water bath. Triethylamine (7.4 mL, 0.053 mol) was added, followed by the dropwise addition of trifluoroacetic anhydride (6.8 mL, 0.049 mol). After 3 h, the mixture was extracted with  $CH_2Cl_2$  (200 mL) and washed with 1 N HCl  $(1 \times 100 \text{ mL})$ , 1 N NaOH  $(1 \times 100 \text{ mL})$ , and water  $(1 \times 100 \text{ mL})$ . The organic layer was dried with MgSO<sub>4</sub>, filtered through celite, and concentrated in vacuo to give 19.49 g (98%) of [4-(3,4 dimethoxyphenyl)-4-(R)-(2,2,2-trifluoroacetylamino)butyl]carbamic acid benzyl ester as a white solid. <sup>1</sup>H NMR  $\delta$  7.31-7.35 (m, 5 H), 6.80-6.89 (m, 3 H), 5.10 (s, 2 H), 4.81-4.93 (m, 1 H), 3.87 (overlapping s, 6 H), 3.13-3.28 (m, 2 H), 1.80-2.00 (m, 2 H),  $1.42-1.59$  (m, 2 H). MS (ES<sup>+</sup>) 455 (M + 1). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient 10-90% (MeCN + 0.16% TFA) in (water  $+$  0.2% TFA) over 4 min,  $t<sub>R</sub>$  = 3.82 min (96%). A 500 mL hydrogenation vessel was charge with this material (19.49 g, 0.043 mol), ethyl acetate (80 mL), ethanol (70 mL), 1 N HCl (20 mL), and 10% palladium on carbon (2.0 g). The mixture was hydrogenated at 50 psig of hydrogen for 24 h, filtered through celite, and concentrated in vacuo to give 13.77 g (90%) of  $N$ -[4-amino-1-(R)-(3,4dimethoxyphenyl)butyl]-2,2,2-trifluoroacetamide as a white solid. <sup>1</sup>H NMR  $\delta$  7.29–7.33 (m, 5 H), 6.81–6.86 (m, 3 H), 5.07 (s, 2 H), 4.80-4.89 (m, 1 H), 3.86 (overlapping s, 6 H), 3.15-3.25  $(m, 2 H)$ , 1.35-1.64  $(m, 6 H)$ . MS (ES<sup>+</sup>) 321 (M + 1). A 500 mL round-bottom flask containing this material (16.85 g, 0.047 mol),  $CH_2Cl_2$  (220 mL), and triethylamine (14.0 mL, 0.10 mol) was cooled using an ice-water bath. Di-tert-butyl dicarbonate (9.77 g, 0.045 mol) was added in one portion. The mixture was stirred at room temperature for 18 h, diluted with  $CH_2Cl_2(300 \text{ mL})$ , and washed with  $1 \text{ N } HCl$  ( $1 \times 100 \text{ mL}$ ),  $1 \text{ N }$ NaOH ( $1 \times 100$  mL), and water ( $1 \times 100$  mL). The organic layer was dried (MgSO4), filtered through celite, and concentrated in vacuo to give  $12.78$  g  $(64\%)$  of  $[4-(3,4\textrm{-dimethoxophenyl})$ 4-(R)-(2,2,2-trifluoroacetylamino)butyl]carbamic acid tert-butyl ester as a white solid. <sup>1</sup>H NMR  $\delta$  6.81 – 6.85 (m, 2 H), 6.78 (s, 1 H), 4.86-4.94 (m, 1 H), 3.89 (s, 3 H), 3.87 (s, 1 H), 3.09-3.24 (m, 2 H), 1.82-2.00 (m, 2 H), 1.47-1.57 (m, 2 H), 1.44 (s, 9 H). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient  $10-90\%$  (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R$  = 3.95 min (99%). A 500 mL round-bottom flask was charged with  $[4-(3,4\t{dimethoxyphenyl})-4-(R)-(2,2,2\t{triffluoro- $\cdot$$ acetylamino)butyl]carbamic acid tert-butyl ester (12.78 g, 0.030 mol), tetrahydrofuran (150 mL), methanol (40 mL), and 3 N NaOH (30 mL). After 3 h, the mixture was diluted with  $CH_2Cl_2$  (500 mL) and washed with water (1  $\times$  100 mL). The organic layer was dried with MgSO4, filtered through celite, and concentrated in vacuo. The crude material was purified via flash silica gel chromatography (230-400 mesh silica gel 60, gradient 90:10 $-40:60$  hexanes: ethyl acetate) to give 9.73 g (99%) of XVI as a white solid. <sup>1</sup>H NMR  $\delta$  6.87 (s, 1 H), 6.82–6.84 (m, 2 H), 3.84-3.90 (m, 7 H), 3.08-3.14 (m, 2 H), 1.62-1.71 (m, 4 H), 1.43 (s, 9 H). MS ( $ES^+$ ) 325 (M + 1). HPLC: Kromasil C18

column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient 10–90% (MeCN + 0.16% TFA) in (water  $+$  0.2% TFA) over 4 min,  $t_R$  = 3.40 min  $(98\%)$ .

Synthesis of N-(2-Thiophenesulfonyl)-{4-(3,4-dimethoxyphenyl)- 4-[4-(4-ethylpiperazin-1-yl)-1-oxo-2,3-dihydro-1H-isoindol-2-yl] butyl}amide (7a). 3-Bromo-2-methylbenzoic acid (15.3 g, 71 mmol) was dissolved in methanol (75 mL) and  $CH_2Cl_2$ (425 mL), cooled to  $-5$  °C, and treated with 2 M trimethylsilyldiazomethane in hexanes (100 mL, 200 mmol) dropwise from an addition funnel. One hour after the addition of reagent, the reaction was complete by LC/MS. Evaporation of volatiles afforded methyl 3-bromo-2-methylbenzoate (16.6 g) as an oil. <sup>1</sup>H NMR  $\delta$  7.71 (t, J = 8 Hz, 2 H), 7.10 (d, J = 8 Hz, 1 H), 3.90 (s, 3 H), 2.63 (s, 3 H). HPLC: Kromasil C18 column (50 mm  $\times$  $2 \text{ mm}$ ;  $3.5 \mu$ ), gradient  $10 - 90\%$  (MeCN + 0.16% TFA) in (water + 0.2% TFA) over 4 min,  $t_R$  = 4.25 (98%). This ester (16.6 g, 71 mmol), N-bromosuccinimide (13.09 g, 74 mmol), and benzoyl peroxide (0.56 g, 2.3 mmol) were dissolved in carbon tetrachloride (180 mL) and heated at 82  $\rm{^{\circ}C}$  (bath) overnight. The reaction was cooled to 23 °C, diluted with ethyl acetate (600 mL), washed with water (300 mL), dried ( $MgSO<sub>4</sub>$ ), filtered, and concentrated in vacuo to afford XVIII as a solid (20.7 g, 95%), which was stored under argon in a freezer. <sup>1</sup>H NMR  $\delta$  7.89 (dd, J = 7.8 and 1.3 Hz, 1 H), 7.76 (dd,  $J=8.0$  and 1.3 Hz, 1 H), 7.23 (t,  $J=8.0$  Hz, 1 H), 5.13 (s, 2 H), 3.96 (s, 3 H). HPLC: Kromasil C18 column  $(50 \text{ mm} \times 2 \text{ mm}; 3.5 \mu)$ , gradient  $10-90\%$  (MeCN  $+ 0.16\%$ TFA) in (water  $+$  0.2% TFA) over 4 min,  $t_R$  = 4.16 min (99%).  $(R)$ -XVI (3.04 g, 8.5 mmol) and triethylamine (1.2 mL, 8.6 mmol) were combined in toluene (50 mL) and treated with XVIII (2.4 g, 7.8 mmol) in toluene (100 mL) via an addition funnel. The reaction mixture was stirred at 23 °C for 1 h, refluxed for 5 h, and evaporated. Purification on silica gel with an Analogix system (SF40-150 g, gradient elution with  $0-2\%$ methanol in  $CH_2Cl_2$ ) afforded XIX (3.9 g, 82%) as a solid. <sup>1</sup>H NMR δ 7.79 (d, J=8.0 Hz, 1 H), 7.39-7.49 (m, 2 H), 6.84-6.98  $(m, 3 H)$ , 5.53 (t,  $J = 7.9$  Hz, 1 H), 4.26 and 3.96 (pair of d,  $J =$ 17 Hz, 1 H each), 3.87 (s, 3 H), 3.85 (s, 3 H), 3.18-3.25 (m, 2 H), 2.08-2.16 (m, 2 H), 1.50-1.60 (m, 2 H), 1.42 (s, 9 H). LC/MS  $(ES+) m/z$  520 (M + 1). HPLC: Kromasil C18 column (50 mm  $\times$ 2 mm; 3.5  $\mu$ ), gradient 10-90% (MeCN + 0.16% TFA) in (water  $+$  0.2% TFA) over 4 min,  $t_R$  = 4.07 min (97%). Compound XIX (3.30 g, 6.0 mmol) was suspended in dry toluene (20 mL) under argon and treated with bis-(tri-tert-butylphosphine)palladium(0) (307 mg, 0.60 mmol), cetyl trimethylammonium bromide (120 mg, 0.33 mmol), and N-ethylpiperazine (2.75 g, 24 mmol). After adding 45% aqueous KOH solution (1.12 g, 9 mmol), the reaction mixture was purged well with argon, sealed with a Teflon screw cap, and heated to 105  $\degree$ C (bath) for 2 h. Analysis by LC/MS indicated 50% XX and 33% des-bromo compound. The reaction mixture was cooled to 23  $\degree$ C, decanted from solids, washed with additional toluene (25 mL), and evaporated in vacuo to afford a brown oil, which was dissolved in  $CH_2Cl_2$  (200 mL), washed with saturated sodium bicarbonate (50 mL), water (50 mL), and brine (50 mL), dried (Na2SO4), and concentrated in vacuo to obtain a solid. The solid was dissolved in ethyl acetate, treated with silica gel (10 g), evaporated, and loaded onto a prepared column of silica gel (2 in. diameter, 200 g, 40:4:0.5 ethyl acetate/methanol/ ammonium hydroxide) and eluted with the same. The productcontaining fractions were evaporated, dissolved in ethyl acetate, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated to give  $XX$  as a foamy solid  $(1.46 \,\text{g}, 44\,\text{V})$ . <sup>1</sup>H NMR  $\delta$  7.51 (d, J = 7.1 Hz, 1 H), 7.37–7.43 (m, 1 H), 7.08 (d, J=7.4 Hz, 1 H), 6.82-6.96 (m, 3 H), 5.53 (t, J=7.8 Hz, 1 H), 4.22 and 3.92 (pair of d,  $J=17$  Hz, 1 H each), 3.87 (s, 3 H), 3.84 (s, 3 H), 3.18-3.24 (m, 2H), 3.05-3.12 (m, 4 H), 2.52- 2.61 (m, 4 H), 2.05-2.15 (m, 2 H), 1.51-1.63 (m, 2 H), 1.42 (s, 9 H). LC/MS (ES+)  $m/z$  553 (M + 1). HPLC: Kromasil C18 column (50 mm  $\times$  2 mm; 3.5  $\mu$ ), gradient 10–90% (MeCN + 0.16% TFA) in (water  $+$  0.2% TFA) over 4 min,  $t_R$  = 3.15 min  $(96\%)$ . A 50 mL round-bottom flask was charged with **XX** (1.63)

g, 2.95 mmol) and 1,4-dioxane (15.0 mL). A solution of 4 N HCl in 1,4-dioxane (10 mL) was added, and the mixture was stirred at 23  $\degree$ C for 4 h. The mixture was concentrated in vacuo to give solid (770 mg, 1.7 mmol), which was dissolved in  $CH_2Cl_2(9 \text{ mL})$ , cooled to 5  $\degree$ C, and treated with triethylamine (0.8 mL, 6 mmol) and 2-thiophenesulfonyl chloride (365 mg, 2.0 mmol) for 1 h. The reaction mixture was diluted with  $CH_2Cl_2(200 \text{ mL})$ , washed with 1 N HCl (50 mL), washed with saturated sodium bicarbonate (50 mL), dried ( $MgSO<sub>4</sub>$ ), and concentrated in vacuo to give an oil that was dissolved in  $CH_2Cl_2$  (20 mL), treated with 1 N anhydrous HCl in diethyl ether, and concentrated in vacuo overnight to afford  $7a$  as a solid HCl salt (0.90 g, 88%). <sup>1</sup>H NMR  $\delta$  7.40-7.57 (m, 3 H), 7.37 (t, J = 7 Hz, 1 H), 7.06 (d, J = 8 Hz, 1 H), 7.00 (t, J=4 Hz, 1 H), 6.80-6.92 (m, 3 H), 5.68 (broad s, 1 H, NH), 5.49 (t,  $J=8$  Hz, 1 H), 4.22 and 3.90 (pair of d,  $J=$ 17 Hz, 1 H each), 3.86 (s, 3 H), 3.82 (s, 3 H), 3.0 (m, 6 H), 2.61 (m, 4 H), 2.50 (m, 2 H), 2.16 (q, J=7 Hz, 2 H), 1.59 (m, 2 H), 1.12 (t,  $J=7$  Hz, 3 H). LC/MS (ES+)  $m/z$  599 (M + 1). Anal. calcd for  $C_{30}H_{38}N_4O_5S_2 \cdot 1.8HCl \cdot 0.5H_2O$ : C, 53.51; H, 6.11; N, 8.32; Cl, 9.48; H2O, 1.34%. Found: C, 53.27; H, 6.04; N, 7.95; Cl, 9.44; H2O, 1.46%. Compound 7a was determined to be 96% ee [Daicel Chiralpak AD-H 25 cm  $\times$  0.46 cm, heptane:2-propanol:Et<sub>2</sub>NH (70:30:0.1), 0.7 mL/min,  $t_R = 20.6$  min; its enantiomer had  $t<sub>R</sub> = 12.2$  min].

N-(2-Thiophenecarbonyl)-{4-(3,4-dimethoxyphenyl)-4-[4-(4 ethylpiperazin-1-yl)-1-oxo-2,3-dihydro-1H-isoindol-2-yl]butyl} amide (7c). The crude BOC-deprotected solid (30 mg, 0.044 mmol) mentioned above (from  $XX$  in preparation of  $7a$ ) was dissolved in DMF (2 mL) under nitrogen and treated with N-methylmorpholine (0.022 mL, 0.20 mmol), 1-methylimidazole-2-carboxylic acid (10 mg, 0.079 mmol), 1-hydroxybenzotriazole (HOBt; 4 mg, 0.03 mmol), O-benzotriazol-1-yl-N,  $N, N', N'$ -tetramethyluronium (HBTU) hexafluorophosphate (30 mg, 0.079 mmol) at 23  $\degree$ C overnight. The reaction mixture was diluted with water (4 mL) and acetonitrile (3 mL), filtered, and purified by reversed-phase HPLC [Phenomenex, Kromasil C18, 5  $\mu$ , 100 Å, 100 mm  $\times$  21 mm, gradient elution with 10-50% acetonitrile  $(0.16\%$  TFA) in water  $(0.2\%$  TFA)] to yield 7c (39 mg, quantitative). <sup>1</sup>H NMR  $\delta$  7.36-7.59 (m, 4 H), 6.81-7.09 (m, 5 H), 5.46–5.51 (m, 1 H), 4.22 and 3.88 (pair of d,  $J=$ 17.9 and 17 Hz, 1 H each), 3.87 (s, 3 H), 3.84 (s, 3 H), 3.49-3.60  $(m, 4 H), 3.21 - 3.37$   $(m, 5 H), 2.97 - 3.12$   $(m, 2 H), 2.17 - 2.28$  $(m, 2H), 1.72-1.85 (m, 2H), 1.52-1.58 (m, 3H)$ . MS (ES<sup>+</sup>) 563  $(M + 1)$ . Anal. calcd for  $C_{31}H_{38}N_4O_6S \cdot 2.1HCl \cdot 1.0H_2O$ : C, 56.65; H, 6.46; N, 8.52; Cl, 11.33; H2O, 2.74. Found: C, 56.30; H, 6.45; N, 8.34; Cl, 11.22; H<sub>2</sub>O, 2.57.

Calcium Flux Assay. A calcium mobilization assay based on a Fluorescence Imaging Plate Reader (FLIPR; Molecular Devices, Sunnyvale, CA) was used to determine the agonist activity of the peptides in CHO cells transfected with rat GPR14 (U-II receptor).<sup>39</sup> To derive these cells, the complete coding sequence of rat U-II (Genbank Accession no. U32673) was amplified by nested PCR from rat heart marathon-ready cDNA. PCR was carried out by using the DNA polymerase PFU (Stratagene) following conditions suggested by the manufacturer. The PCR products were cloned into pcDNA3 (Invitrogen) digested with EcoR I and Xba I. Clones containing rat UT were verified by complete sequencing of the U-II receptor insert to ensure a lack of PCR-introduced errors. The constructed vector was transfected into CHO cells by using lipofectamine (GIBCO BRL). CHO cells with high expression of rat UT (determined by response to U-II in FLIPR-based calcium mobilization) were selected and established as stable cell lines by using G418. CHO cells were seeded at 40000 cells per well into 96-well, black-wall, clear-bottom microtiter plates 24 h before running the assay. Cells in culture media (DMEM/F12 containing 15 mM HEPES, L-glutamine, pyridoxine hydrochloride, 10% fetal bovine serum, 1 mg/mL G418 sulfate, antibiotic/antimycotic, pH 7.4) were loaded with proprietary dye, from the FLIPR calcium assay kit (Molecular Devices),

prepared in assay buffer (Hanks Balanced Salts Solution, 20 mM HEPES, 0.1% bovine serum albumin (BSA), 2.5 mM probenecid, pH 7.4), and incubated for 1 h at 37 °C. Calcium mobilization determinations were performed at 23 °C.  $EC_{50}$ values were calculated relative to the maximum calcium ionophore (A23187) response at 10  $\mu$ M. Antagonist activity of inactive compounds was assessed, after a 10 min of incubation, in response to an agonist, such as goby U-II (0.1 nM) or peptide 1 (0.3 or 1 nM). The use of rat GPR14 and the goby peptide was considered acceptable because human and goby U-II have similar affinity for human or rat GPR14 in the transfected cells.<sup>6</sup>

Radioligand Binding Assay.<sup>24</sup> CHO cells stably expressing rat  $GPR14^{39}$  (UT) were harvested and washed with ice-cold phosphate buffered saline (PBS, pH 7.4) and resuspended in 20 mM Tris-HCl (pH 7.4), 250 mM sucrose, and 1 mM ethyleneglycol tetraacetic acid. Cells were homogenized in a glass-Teflon homogenizer and, after centrifugation at 1000g for 15 min, the resulting supernatant was obtained and recentrifuged at 30000g for 60 min at 4  $\degree$ C. The resulting pellet was passed through a 25 gauge needle five times, and membranes were stored at  $-80^{\circ}$ C. Membranes were precoupled to wheat germ agglutinin-coated scintillation-proximity assay (SPA) beads (Amersham Pharmacia Biotech, England, UK) at the concentration of 15  $\mu$ g membrane/mg beads. The coupling was carried out for 30 min at room temperature in buffer A (50 mM Tris-HCl, 5 mM  $MgCl<sub>2</sub>$ , and 0.1% BSA adjusted to pH 7.4) with continual shaking. After a brief spin, the supernatant was discarded. The assay was conducted in 96-well microtiter plates, with a final concentration of 1.0 mg of precoupled beads per well, 0.17 nM<br>monoiododinated <sup>125</sup>I-[Tyr]-rat U-II (Phoenix Pharmaceuticals, Belmont, CA), and various concentrations of compounds in a total volume of 100  $\mu$ L buffer A. The plate was shaken for  $3 h$  at  $22 °C$ , and after a brief spin, the radioactivity was counted by using a Topcount instrument (Packard Bioscience Company, Meriden, CT). Nonspecific binding, determined using  $1 \mu M$  rat U-II, was always less than 15% of the total binding. The GraphPad Prism software package (GraphPad Software, Inc., San Diego, CA) was used to analyze the binding data, and  $K_i$ values were calculated  $(n=2)$  by the Cheng-Prussoff equation.

Functional Assay in Rat Aortic Rings.<sup>27</sup> Selected compounds were tested for contractile effects on rat thoracic aorta smooth muscle at a concentration of 30 nM. The response  $(n=2)$  was measured relative to that produced by 80 mM KCl. These studies were conducted by CEREP (Celle L'Evescault, France). The aortas were obtained from male Wistar rats (250 g; CERJ, Le Genest St. Isle, France). The animals were sacrificed by cervical dislocation followed by exsanguination, and the aortas were quickly removed, cleaned, and dissected into rings. The rings were suspended between two stainless steel hooks in organ baths containing 20 mL of an oxygenated and prewarmed physiological salt solution. They were connected to force-displacement transducers (UF1, Phymep) for isometric tension recordings. Data were continuously recorded on a multichannel data acquisition system. The physiological salt solution contained the following (mM): NaCl, 118.0; KCl, 4.7;  $MgSO<sub>4</sub>$ , 1.2; CaCl<sub>2</sub>, 2.5; KH<sub>2</sub>PO<sub>4</sub>, 1.2; NaHCO<sub>3</sub>, 25.0; glucose, 11.0. The high- $\bar{K}$ <sup>+</sup> (80 mM) solutions were prepared by equimolar replacement of NaCl with KCl. These solutions were continuously aerated with a 95%  $O_2/5\%$   $CO_2$  gas mixture and maintained at pH 7.4 and 37 °C. Goby U-II (Palomar Research, Inc., cat. no. U4832, lot no. A34001-298M) and test compounds were freshly prepared in stock solutions by using physiological buffer. The aortic rings were stretched to an optimal resting tension and then allowed to equilibrate for at least 30 min, during which time they were washed repeatedly. Each ring was contracted using a high KCl solution (80 mM) to verify responsiveness and to obtain a maximal contractile response. The rings were again washed until resting tension returned to baseline and then they were exposed to 30 nM of the test compound

 $(n=2/\text{test compound})$  until a stable response was obtained or for a maximum of 30 min. The degree of contraction was expressed as a percentage of the KCl-induced maximal contraction.

Acknowledgment. We thank Benôit Liégault for initial development of the palladium-catalyzed aryl amination reaction and Andrew Mahan for chiral HPLC analyses.

Supporting Information Available: Characterization data for compounds 2,  $(R)$ -4,  $(S)$ -4,  $5a-g$ ,  $5k-m$ ,  $(R,R)$ -5a,  $(R)$ -5j,  $(S)$ -5j, 6a-j, 7b, 7d-o. Energy-minimized structures for 7a and 8-12. This material is available free of charge via the Internet at http://pubs.acs.org.

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